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Unit C9 Chemistry of the atmosphere

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The	The composition and evolution of the Earth's atmosphere			16	Human activities	deforestation, burning fossil fuels, decomposition of			
1	Approximate	Nitrogen	79%			that increase CO ₂	waste in landfills		
	proportions of gases	Oxygen	21%		17	Human activities	agriculture: animal digestion, rice farming. decompos		rming. decomposition
	in the atmosphere	Carbon	0.04%			that increase CH ₄	of waste		
		dioxide			18	Effects of climate	melting polar ice caps, changes in rainfall patterns, mo		infall patterns, more
		Other gases	Include variable quantities			change	frequent stor	frequent storms, temperature increase	
			water vapour and noble gases		The	carbon footprint and it	s reduction		
2	The early atmosphere		nosphere 4.6 million years ago		19	Carbon footprint	total amount of carbon dioxide and other greenhouse		
3	Theories about the	•	here of mars, mainly carbon diox	ide,			gases emitted during the lifetime of a product		
	early atmosphere	little or no oxyg			20	Reducing the carbon	government tax on greenhouse gas emissions,		
4	The first billion years	•	released water vapour and nitro	gen		footprint	government sell industry licences, carbon capture		•
5	Formation of the	water vapour fr	om the atmosphere condensed				technology, reduce use of non-renewable fuels		
	oceans				21	Difficulties reducing	•	conomic growth, chang	ges to lifestyles
6	Reducing carbon		dissolved in the oceans and carbo	onates		carbon footprint	needed, lack of education		
	dioxide	were precipitate					itants and their sources		
7	Green plants		lioxide and release oxygen during	g	22	Atmospheric	harmful particles and gases released into the		d into the atmosphere
		photosynthesis				pollutants			
8	Photosynthesis	Carbon dioxide + Water → Oxygen + Glucose			23	Examples of	carbon dioxide (CO ₂), water vapour (H ₂ O _(g)), carbon		
			$6H_2O \rightarrow 6O_2 + C_6H_{12}O_6$			atmospheric	monoxide (CO), sulfur dioxide (SO ₂), oxides of nitrogen,		
9	Algae	•	existed 2.7 billion years ago			pollutants	particulates		
10	"Locking up" carbon		dimentary rocks and fossil fuels		24	Sources and effects	Pollutant	Source	Effect
	- 11.6	•	on caused further decrease of CO			of atmospheric	$CO_{2,} H_2O_{(g)}$	Combustion of fossil	Greenhouse effect
11	Fossil fuels		years plankton and plant deposit	ts are		pollutants		fuels	F
_			g. coal, oil, natural gas				СО	Incomplete combustion	Fainting, coma, death
	bon dioxide and methane						SO ₂	Combustion of coal	Acid rain (kills
12	Greenhouse gases		yer of gases that maintain Earth's	6				(SO ₂ mixes with	plants, corrodes
4.2	ı		t a level to support life					clouds, producing	buildings and
13	e.g. greenhouse gases		methane and water vapour					sulfuric acid)	metal)
14	Greenhouse effect	-	gases don't absorb incoming sho	ort			Nitrogen	Internal combustion	Acid rain
			diation from the sun				oxides	engines	7.6.6
			ngth radiation that is reflected ba	ack			particulates	Soot from	Respiratory
			is reradiated in all directions					combustion	problems, global
		_	diation is thermal radiation, so						dimming
4-	Climata da com	warms the surface of the Earth		-	25	Incomplete	Not enough o	xygen present for com	plete combustion
15	Climate change	· ·	ture is increasing due to rising CO	\bigcup_2		combustion			
		levels							